

SUPPLEMENTARY MATERIALS

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Magnetically Separable $\text{Fe}_3\text{O}_4/\gamma\text{-Fe}_2\text{O}_3@\text{MIL-88b(Fe)}$ and $\text{Fe}_3\text{O}_4/\gamma\text{-Fe}_2\text{O}_3@\text{NH}_2\text{-MIL-88b(Fe)}$ Composites for the Photocatalytic Degradation of Congo Red Dye

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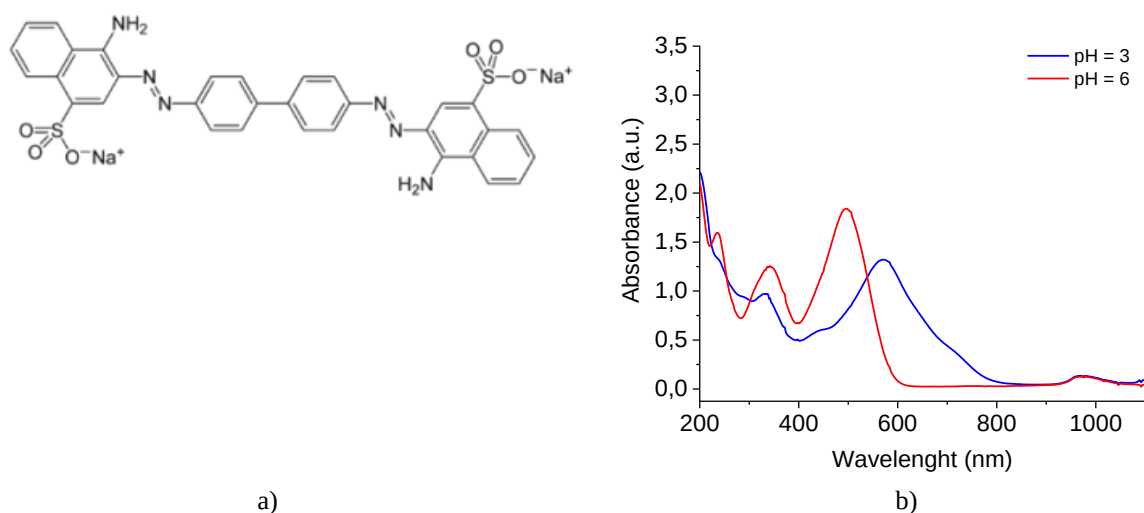


Figure S1. Congo Red structure (a) and UV-Vis spectra at pH of 3 and 6 (b)

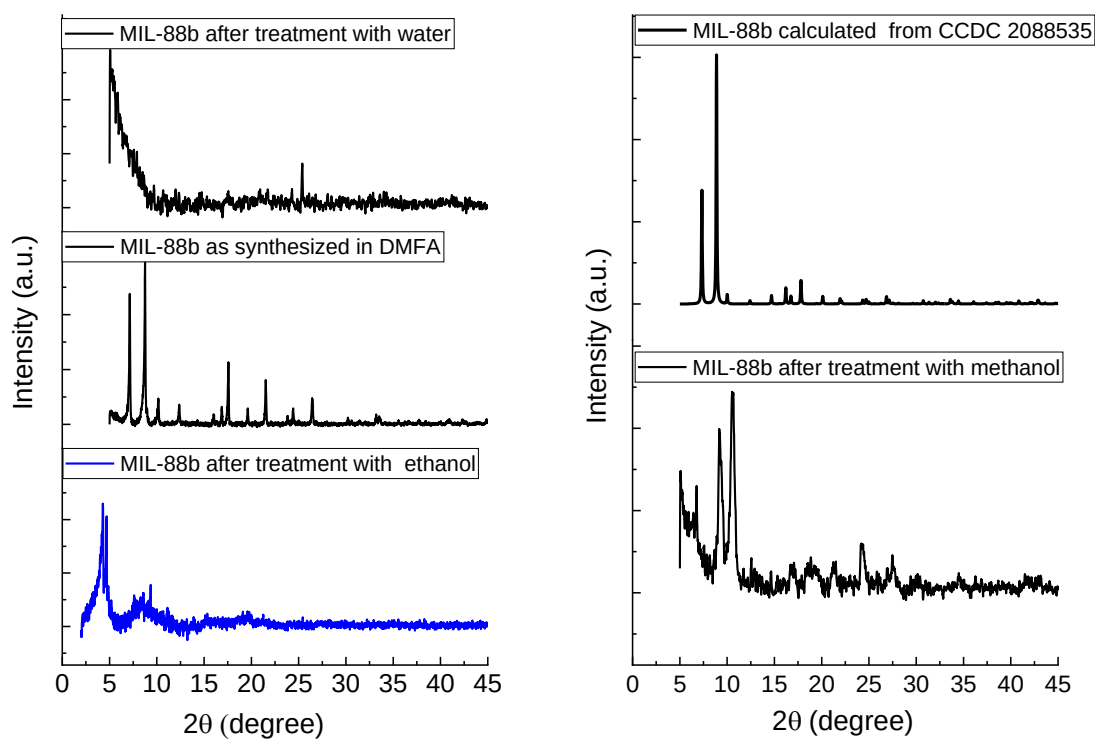


Figure S2. XRD patterns for MIL-88 as synthesized with different guest molecules and the calculated CCDC 2088535 (Cu-K α radiation, $\lambda = 1.5406 \text{ \AA}$)

T a b l e S 1

Removal of organic pollutants using different catalysts

Catalyst	Irradiation source	Reaction conditions	Results	k, min ⁻¹	Ref.
MIL-88A	300 W Xenon lamp	Catalyst=0.25 g/L; [Tetracycline hydrochloride, TC] =100 mg/L; [S ₂ O ₈ ²⁻]=4 mM.	100 % TC was moved after 90 min	0,0396	1
MIL-88A	9W UVA lamp	Catalyst=0.5 g/L; [Rhodamine B, RhB]= 10 mg/L; [S ₂ O ₈ ²⁻]=1 mM.	after 2 hours, the decolorization efficiency reached ~80 %	0,461	2
MIL-88A@ polyacrylate composite	500 W Xe lamp	Catalyst=0.25 g/L; [Rhodamine B, RhB]= 10 mg/L; 20 mM H ₂ O ₂	45 % RhB at 80 min, 97 % after 5 cycles	0,0726	3
MIL-88A@GO	500 W Xe lamp	Catalyst=0.25 g/L, [Rhodamine B, RhB]= 10 mg/L 20 mM H ₂ O ₂	-	0,0645	3
MIL-88A(Fe)	natural sunlight	Catalyst=0,2 g/L; [Rhodamine B. RhB] = 10 mg/L; 20 mM H ₂ O ₂	98 % degradation of RhB in 50 min	0,0793	4
MgFe ₂ O ₄	150 w Xenon lamp	Catalyst=0,8 g/L; [Congo red, CR] = 10 mg/L; pH = 6, [H ₂ O ₂] = 2.5 mM.	removal efficiency of 84.58 %	0.0167	5
FeSO ₄ .7H ₂ O/ H ₂ O ₂	UV lamp	[Fe ²⁺] = 0,4 g/L; [CR] = 30 mg/L, [H ₂ O ₂] = 0,350 g/L; pH=3;	removal efficiency of 79.33 % after 330 min	0.0122	6
TiO ₂	125 W mercury lamp	Catalyst=0,5 g/L [CR] = 60 mg/L, pH=3.5;	removal efficiency of 86 % after 120 min	0.0267	7
Co ₃ O ₄ /TiO ₂ /GO	300W Xenon lamp	Catalyst=0,5 g/L; [CR] = 10 mg/L,	removal efficiency of 91 % after 90 min	0.0262	8
SnO ₂ -Fe ₃ O ₄	14 W UV lamp	Catalyst=1 g/L; [CR] = 18 mg/L; [H ₂ O ₂] = 150 mM.	removal efficiency of 51 % after 90 min	0.024	9
Bentonite-Fe ₃ O ₄ (1:1)	visible irradiation	Catalyst=0.05 g/L; [CR] = 25 mg/L; pH=5; [H ₂ O ₂] = 530 mM	removal efficiency of 98 % after 90 min	0.0348	10
Fe ₃ O ₄ /MIL-88b (Fe)	26W UV lamp	Catalyst=2 g/L; [CR] = 60 mg/L; pH=3.5; [H ₂ O ₂] = 176 mM	removal efficiency of 89 %	0.013±0.001	this work
Fe ₃ O ₄ / NH ₂ -MIL-88b(Fe)	26W UV lamp	Catalyst=2 g/L; [CR] = 60 mg/L; pH=3.5; [H ₂ O ₂] = 176 mM	removal efficiency of 95 %	0.023±0.002	this work

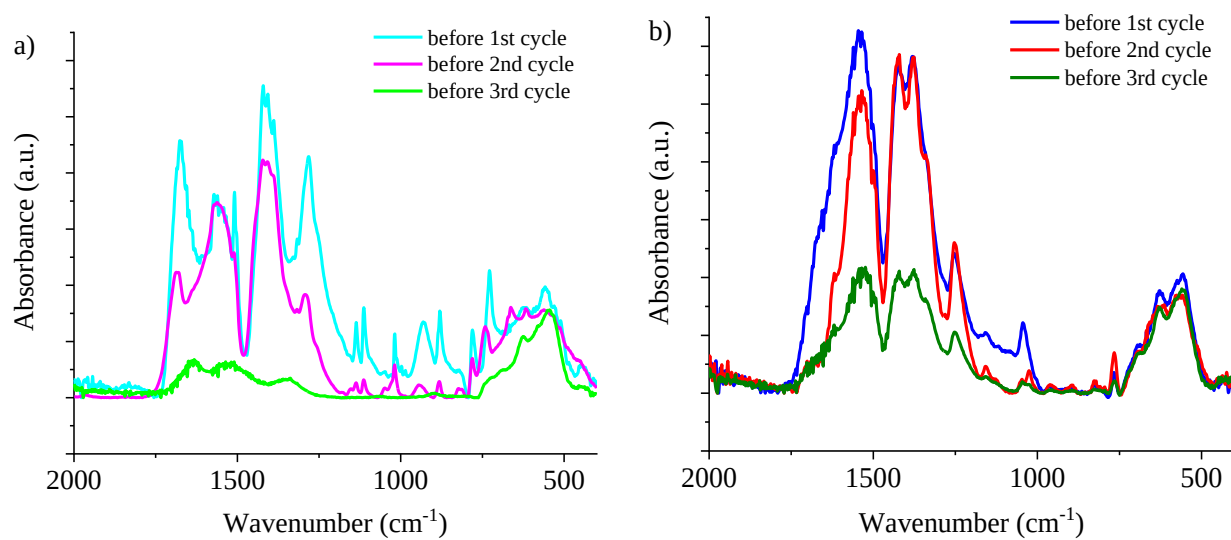


Figure S3. FTIR spectra of Fe₃O₄/γ-Fe₂O₃@MIL-88b(Fe) (a) и Fe₃O₄/γ-Fe₂O₃@NH₂-MIL-88b(Fe) (b) before each cycle

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